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**Content Based
Chemical Engineering**

ENGINEERING DESIGN CALCULATION - DENNIS KIRK ENGINEERING

Calculation of Gas Properties using EOS - (refer Dr Maria A Barrufet, Texas A&M University - March 1988)

Version 1.200

Generalised Cubic Equations Of State:

$$P = \frac{RT}{V-b} - \frac{a}{V(V+b_1)(V+b_2)}$$

$$a = a_c \cdot f(T_r, w)$$

Equations Of State:

- Van der Waals
- Redlich-Kwong
- Soave-Redlich-Kwong
- Peng-Robinson

	b	b1	b2	a _c	a (Tr,w)
VdW	0.125.R.Tc/Pc	0	0	0.42188.R ² .Tc ² /Pc	1
RK	0.08664.R.Tc/Pc	-b	0	0.42748.R ² .Tc ² /Pc	Tr ^{-1/4}
SRK	0.08664.R.Tc/Pc	-b	0	0.42748.R ² .Tc ² /Pc	f _{SRK} (Tr,w)
PR	0.07780.R.Tc/Pc	-b.(1+sqrt(2))	-b.(1-sqrt(2))	0.45724.R ² .Tc ² /Pc	f _{PR} (Tr,w)
	f _{SRK} (Tr,w)=1+(0.480+1.574.w-0.176.w ²). (1-sqrt(Tr))				
	f _{PR} (Tr,w)=1+(0.37464+1.54226.w-0.26992.w ²). (1-sqrt(Tr))				
	VdW	RK	SRK	PR	

Pressure	P	100	kPa(g)
Temperature	T	20	°C
Molecular Weight	MW	18.594	kg/kmol
Critical Temperature	Tc	204.88	°K
Critical Pressure	Pc	4589	kPa(abs)
Acentric Factor	w	0.0248	

	VdW	RK	SRK	PR		
Density	d (kg/m ³)	1.54389391	1.54355086	1.54313259	1.54491799	OK
Molar Vol	V (m ³ /kmol)	12.0435736	12.0462503	12.0495155	12.0355903	= MW/d
Compressibility	Z	0.99478	0.99500108	0.99527078	0.99412059	= P.V/(R.T)
				0.02888		

Pressure	P		psi(g)
Temperature	T		°F
Molecular Weight	MW		lb/lb-mol
Critical Temperature	Tc		°R
Critical Pressure	Pc		psi(a)
Acentric Factor	w		

	VdW	RK	SRK	PR	
Density	d (lb/ft ³)	0.09636986	0.09634844	0.09632234	0.09643378
Molar Vol	V (ft ³ /lb-mol)	192.944146	192.987028	193.039338	192.816249
Compressibility	Z	0.99478	0.99500108	0.99527078	0.99412059